

The Counterflow Virtual Impactor (CVI)

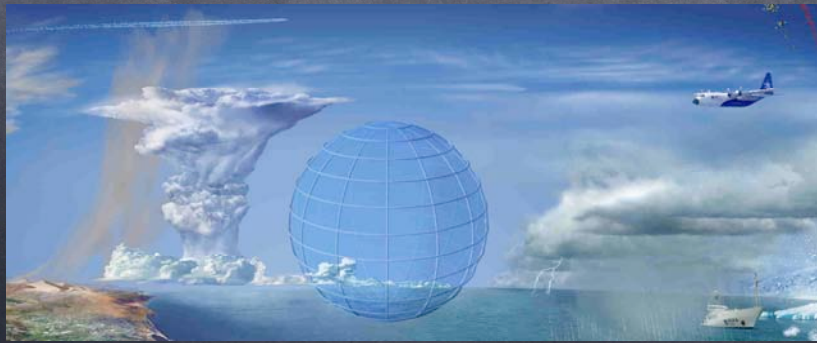
Kevin Noone
Executive Director, IGBP



Outline

- What questions stimulated the development of the CVI?
- Basic principles
- Example results
- Ground-based examples
- Airborne examples

The big question:
Which particles form
droplets and ice crystals,
and which do not?



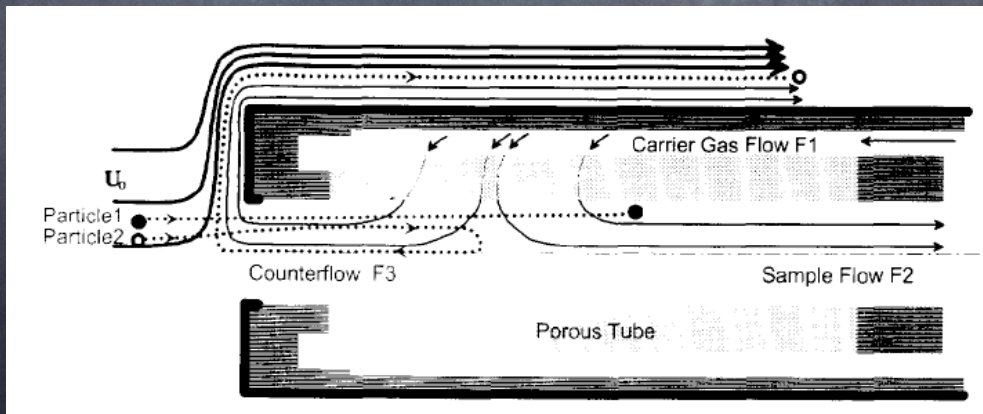
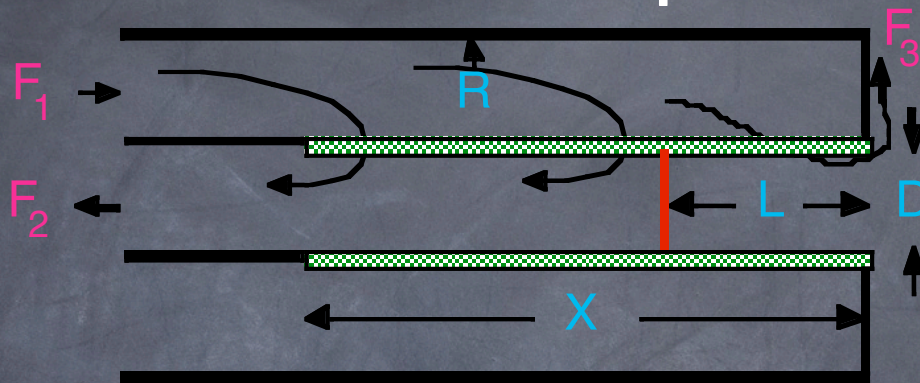
A Very Useful Number

$$Stk = \frac{\tau V}{L} = \frac{\rho_p d_p^2 V}{18\mu L}$$

Small Stokes numbers: particles behave like gases

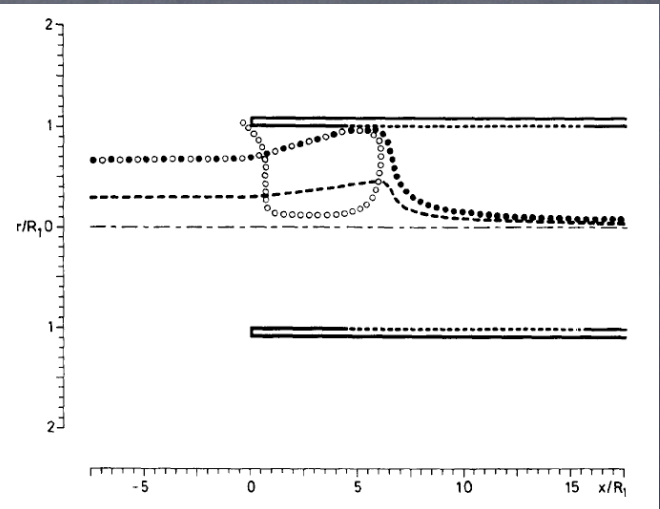
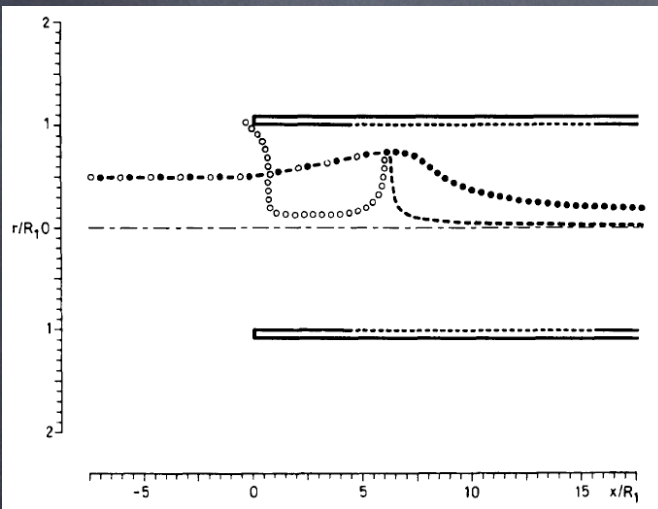
Large Stokes numbers: particles behave like rocks

Basic Principles



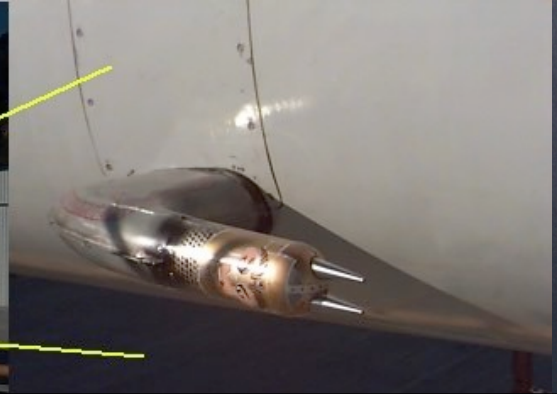
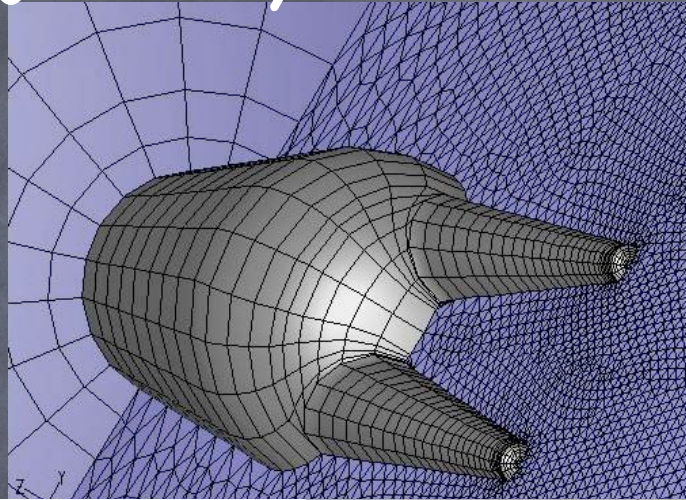
Lin & Heintzenberg, J. Aerosol Sci. 26, 903-914, 1995

Basic Principles

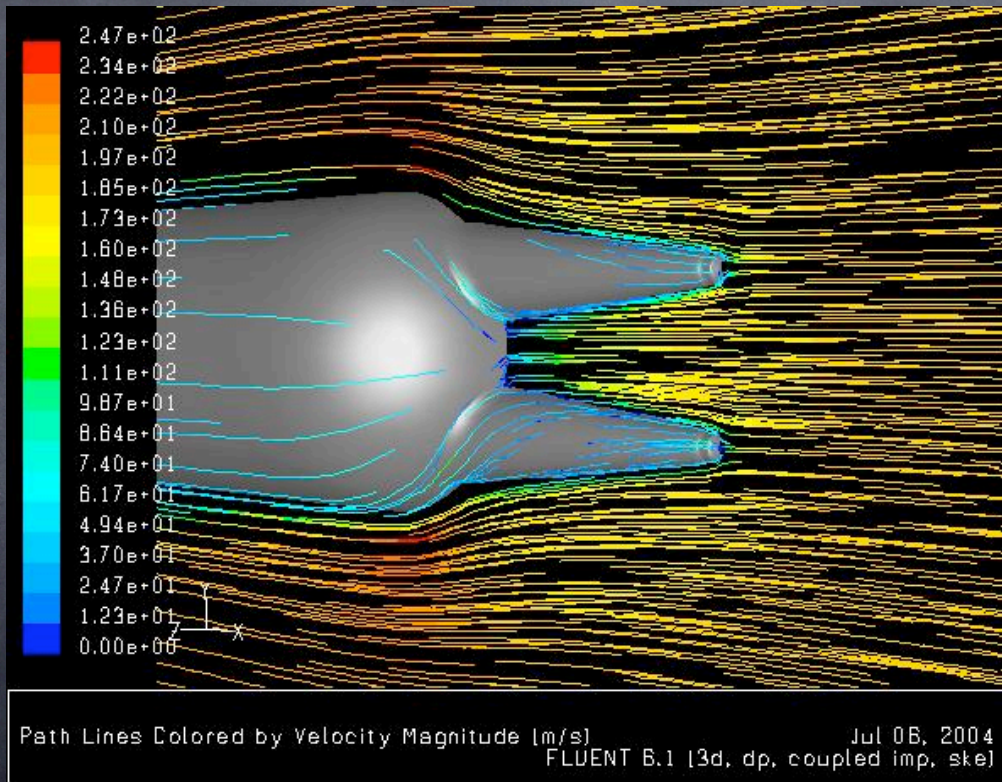


Lin & Heintzenberg, J. Aerosol Sci. 26, 903-914, 1995

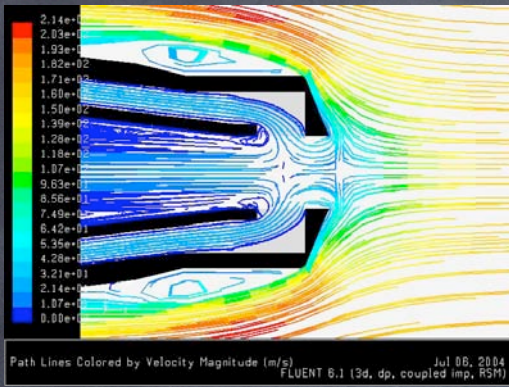
Trajectory calculations



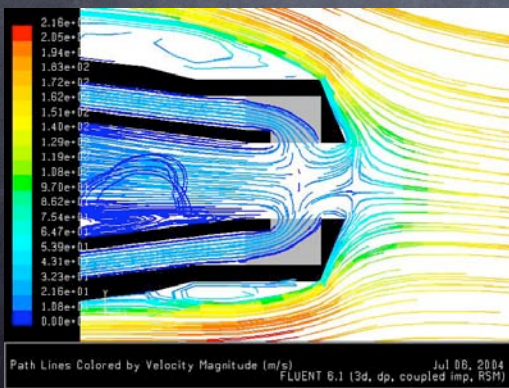
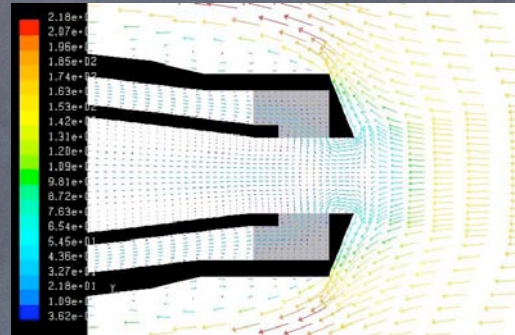
Exterior flow



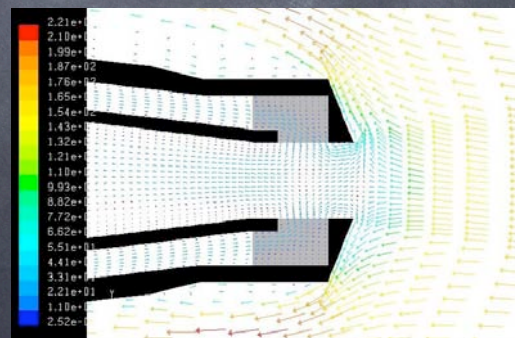
Interior flow



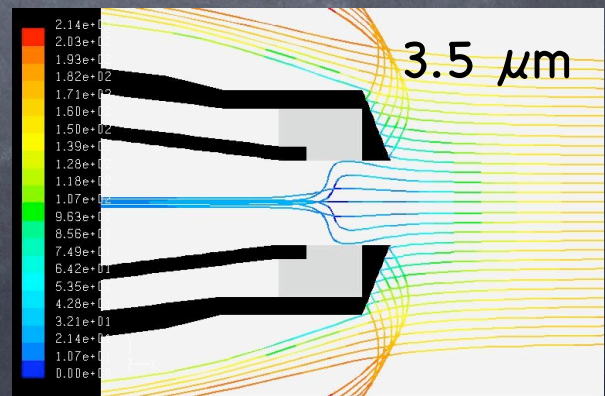
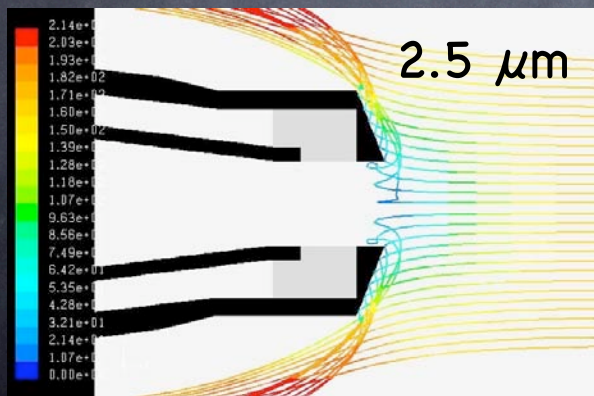
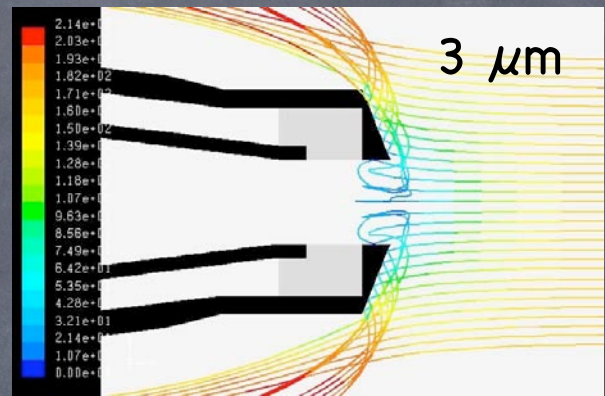
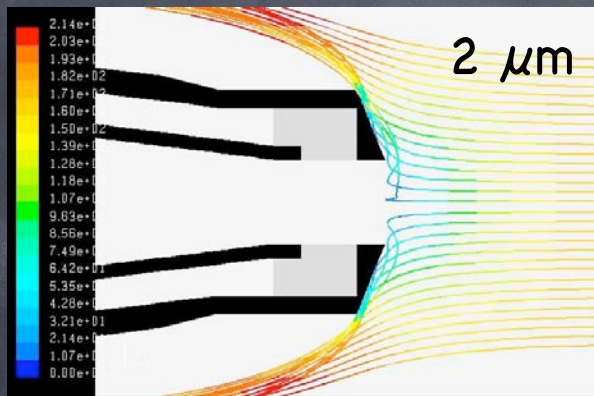
0°



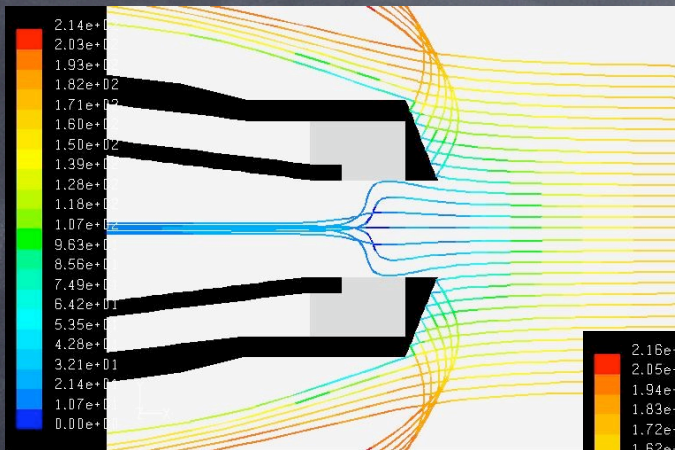
5°



Interior trajectories

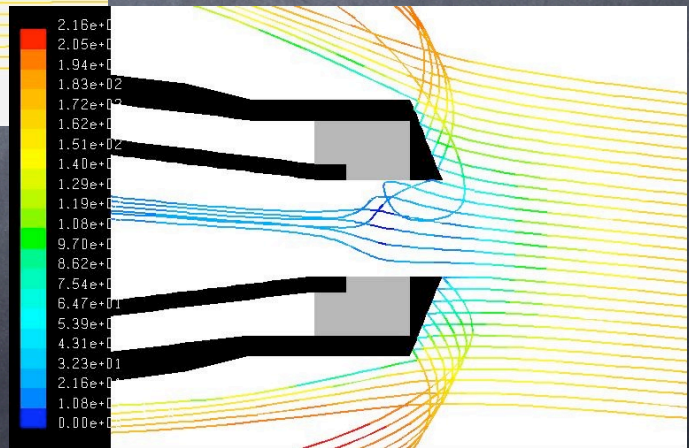


Interior trajectories

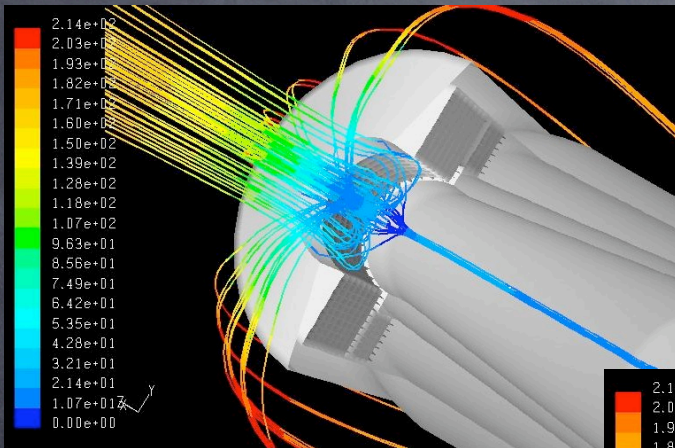


0° attack angle

5° attack angle

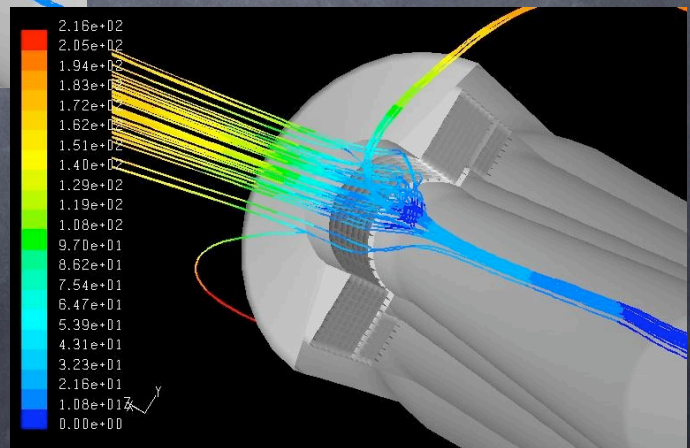


Trajectories

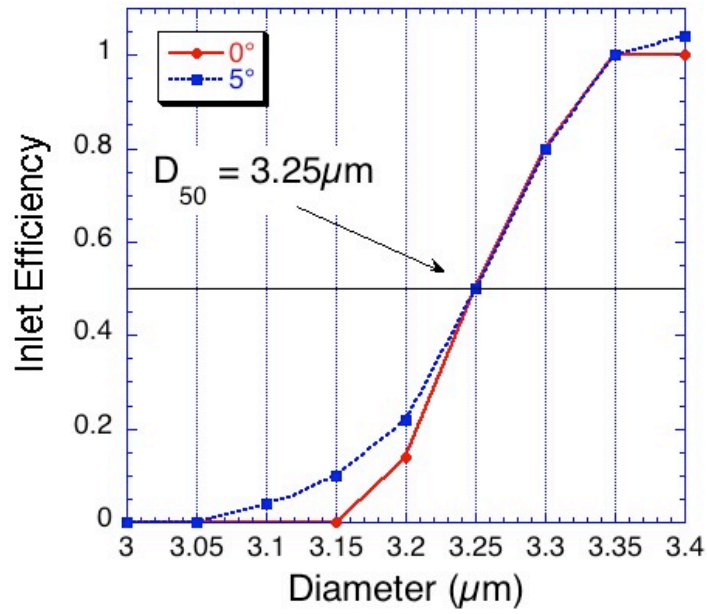


3.5 μm, 0° attack angle

3.5 μm, 5° attack angle

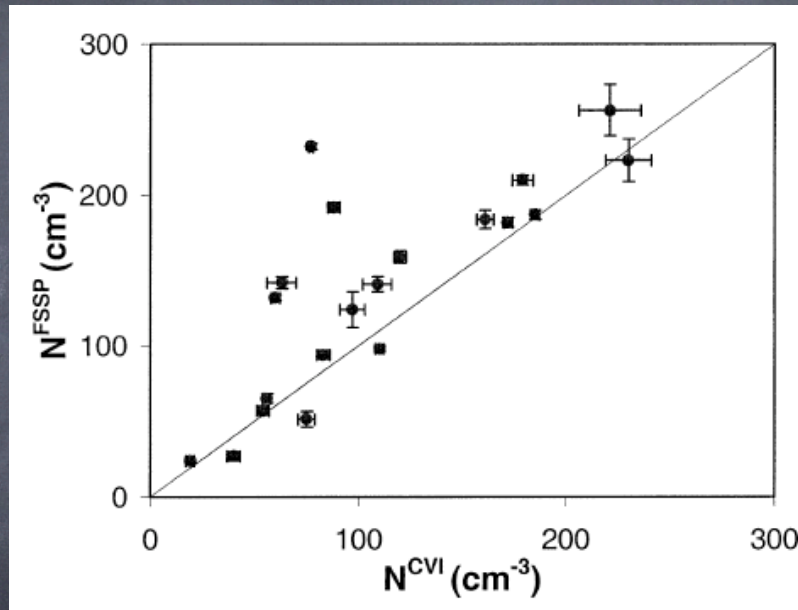


Sampling efficiency



CVI-FSSP Comparison

UK
C-130
+
UW
C-131



Glantz, P., K. J. Noone, and S. R. Osborne, 2003: J. Atmos. Ocean Technol., 20, 133-142.

Example measurements

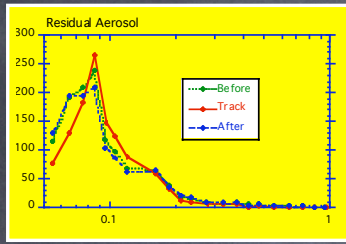
Direct	Indirect
<p>CWC</p> <p>N_{res}</p> <p>SD_{res}</p> <p>$\sigma_{s,res}$</p> <p>$\sigma_{a,res}$</p> <p>Resid. Chem.</p>	<p>M_{res}</p> <p>Mass conc. in droplets</p> <p>Mass avg. D_d</p> <p>Particle size determining N_d</p> <p>Conc., comp. vs size</p>

Airborne examples

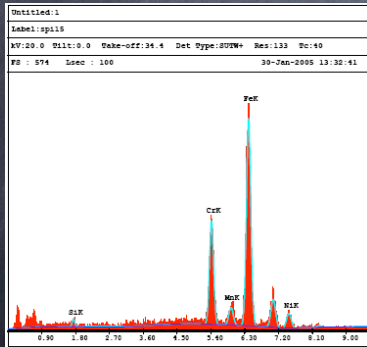
INTACC
INCA



Snoopy ACE-2, INTACC Payload



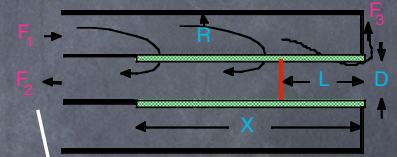
PCASP



Filter/SPA

Condensation Particle Counter (CPC)

Cloud Droplet Residual Particles



Condensed water content (CWC) with Lyman-alpha hygrometer

INTACC chemical analysis

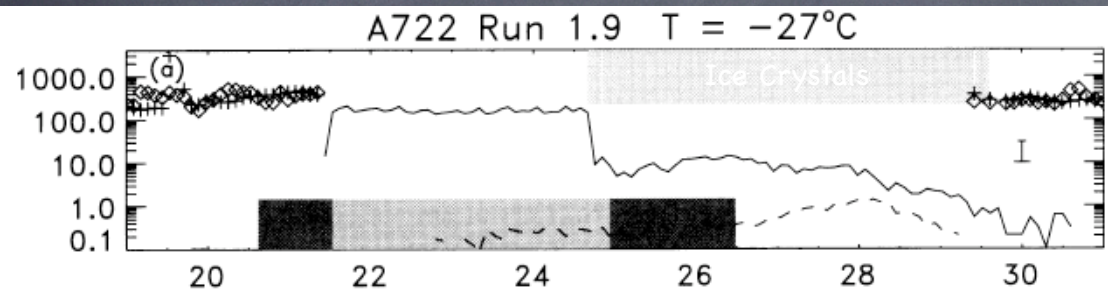
- ★ Ice crystal residuals collected on Nuclepore polycarbonate membrane were analyzed with SEM-EDAX (*Energy dispersive x-ray analysis*)
- ★ Only elements with atomic number $Z \geq 11$ (Na) were considered in this study
- ★ Only particles ≥ 100 nm were analyzed
- ★ Groups identification: Hierarchical cluster analysis was applied to the X-ray intensities

Orographic supercooled clouds

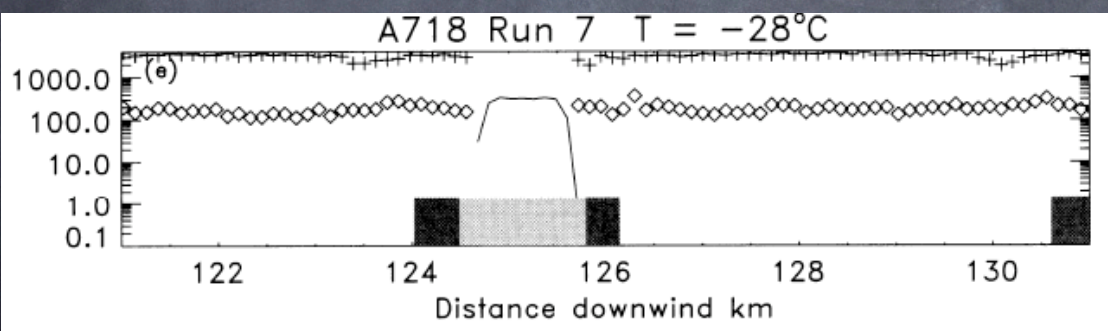


- ★ Low N scavenging fractions ($\leq 10\%$)
- ★ N dominated by submicrometer aerosol

Chemical influences on freezing



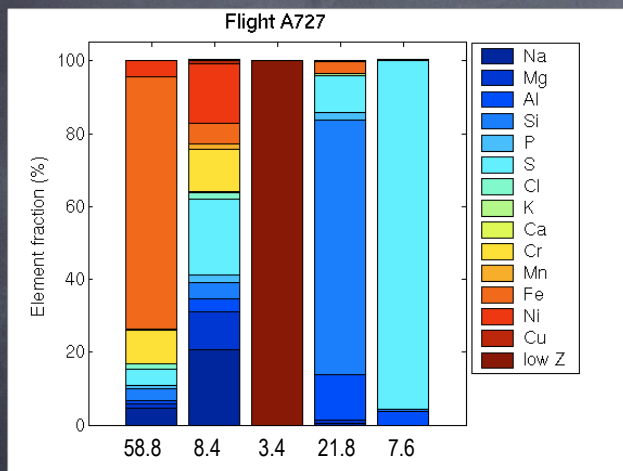
Continental



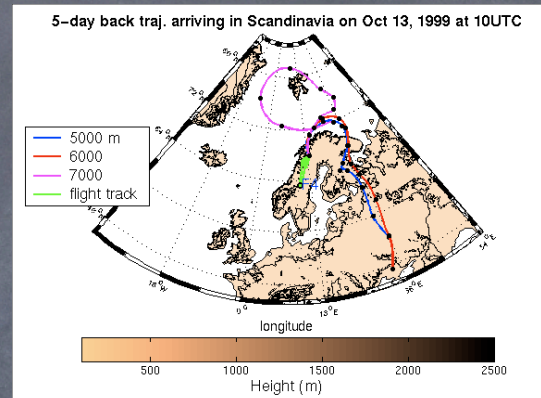
Marine

Field, et al., Q. J. R. Meteorol. Soc., **127**, 1493-1512, 2001

Continental case

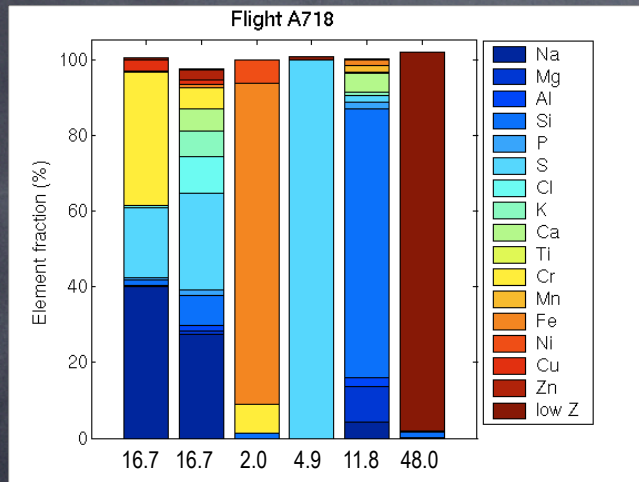


Targino, et al., Atmos.
Phys. Chem. 6, 1977-1990,
2006

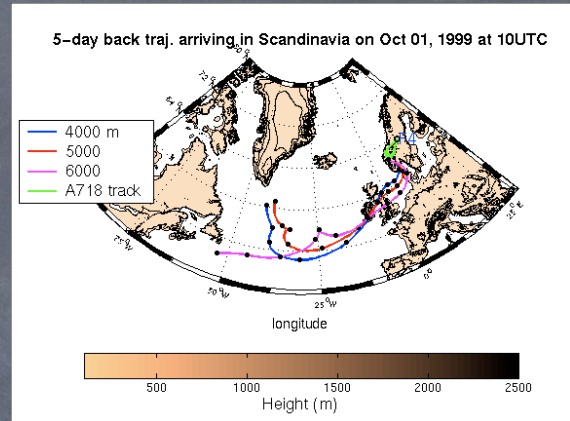


Cr-Fe:	58.8%
Al-Silicates:	21.8
Mixed mineral:	8.4
S + organics:	7.6
Organics:	3.4

Clean marine case

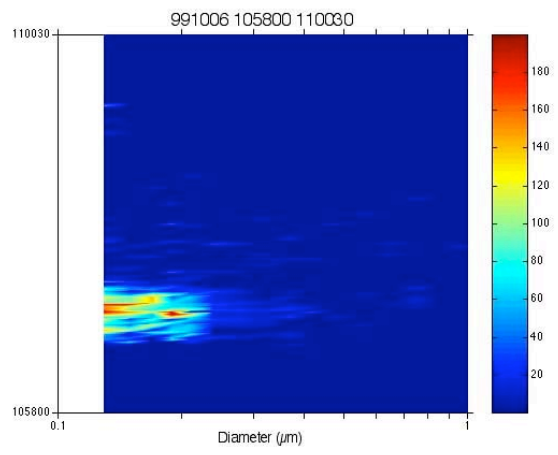
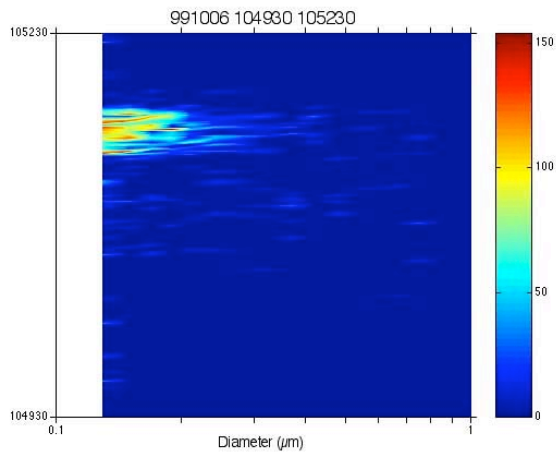
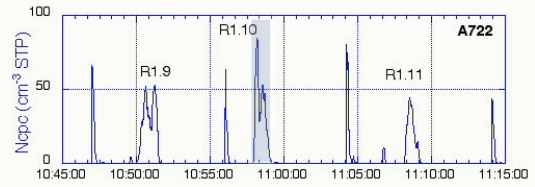
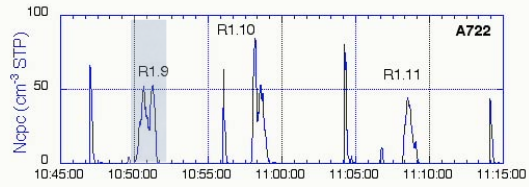


Targino, et al., Atmos.
Phys. Chem. 6, 1977-1990,
2006



Organics:	48.0%
Aged sea salt:	16.7
Aged Cr-Na:	16.7
Silicates:	11.8
S + organics:	4.9
Fe-rich:	2.0

Dilute droplets froze first



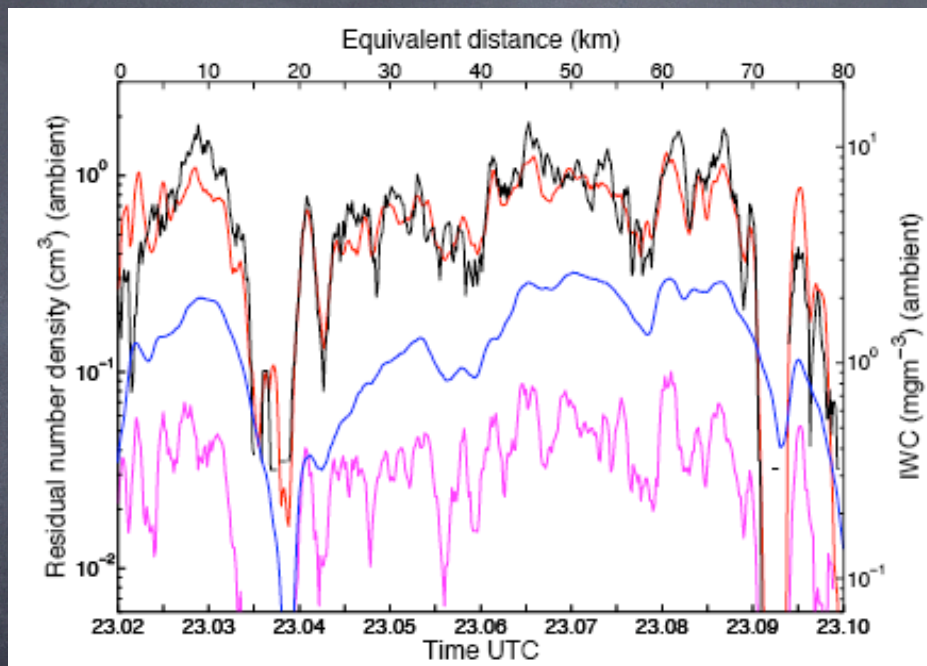
ice ⇌ liquid Alt = 5760m T = -27°C

liquid ⇌ ice Alt = 5590m T = -27°C

INTACC conclusions

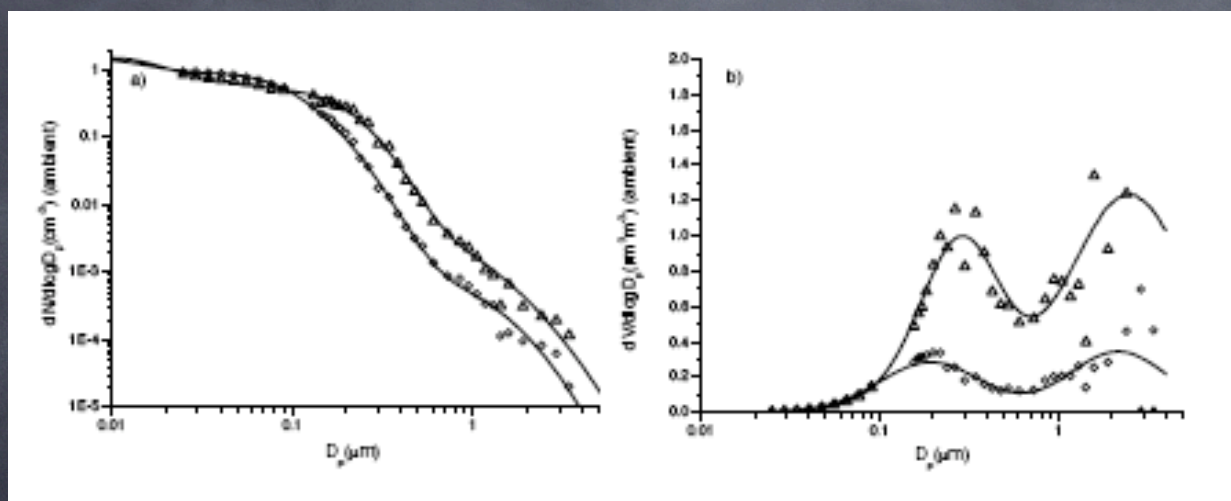
- Freezing occurred very rapidly in the downdraft part of the cloud
- Freezing was inhibited when organic aerosols were present in large concentrations
- More dilute droplets (those that formed on smaller particles) appeared to freeze first

INCA



Seifert, et al., Atmos. Chem. Phys. 2002

INCA



Triangles - SH; Circles - NH

Seifert, et al., Atmos. Chem. Phys. 2002

INCA Conclusions

- SH residuals are larger, and more aerosol volume is incorporated into the crystals than in the NH
- Residual number dominated by particles $< 0.1\mu\text{m}$, volume by particles $> 0.1\mu\text{m}$. Sub- $0.1\mu\text{m}$ particles will control initial crystal concentrations
- The shape of the RSD is insensitive to T variations
- As crystal number increases, the residual volume mean diameter increases (large particles become more important)
- Scavenging ratios of 1% or less, and does not vary much with particle size; no strong preference for nucleation to occur on large particles
- Excellent agreement between CVI and FSSP-300

Seifert, et al., Atmos. Chem. Phys. 2002

Ground-based example

SOACED



- ★ What are the size distributions of scavenged and interstitial aerosol?
- ★ What is the chemical composition of the aerosol?
- ★ What factors control the activation of cloud droplets?



Hypotheses & questions

Accumulation-mode particles control the number of cloud droplets

Organic compounds are a substantial fraction of the aerosol that forms cloud droplets

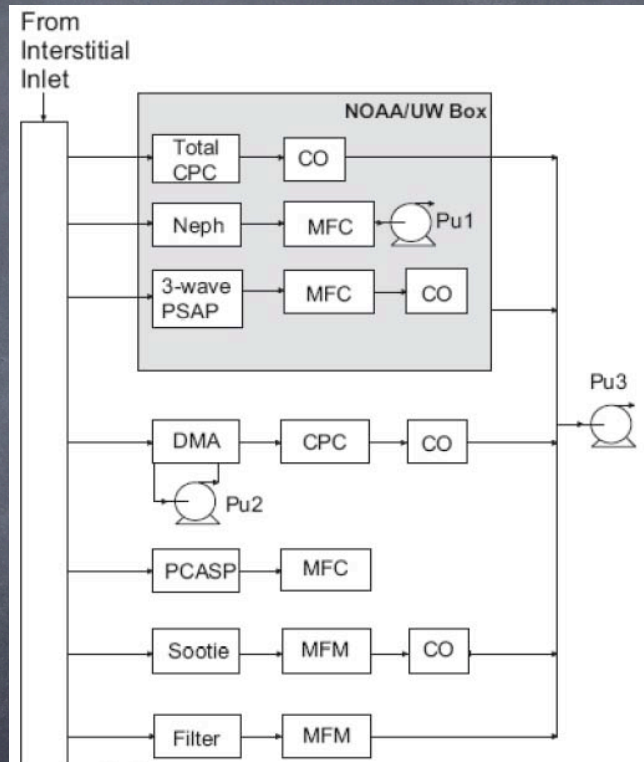
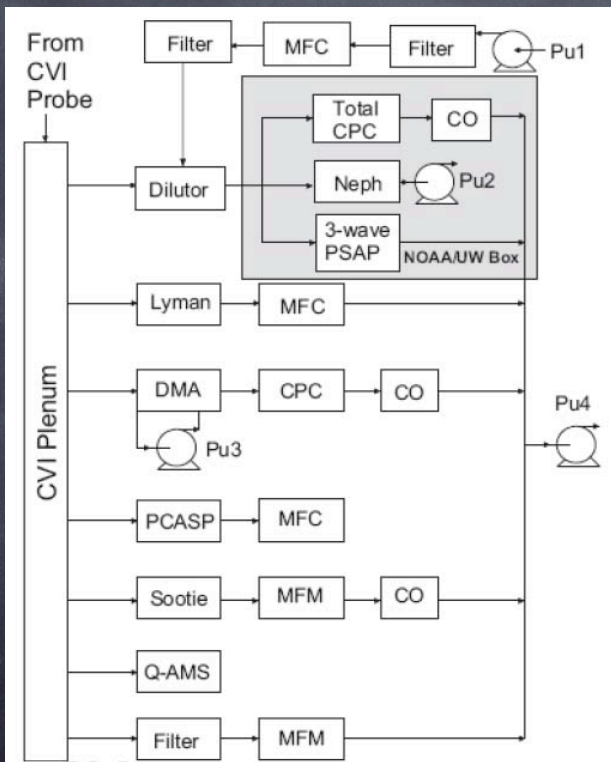
Organic compounds enhance ice formation in supercooled clouds

What are the size distributions of the scavenged and interstitial aerosol?

What is the chemical composition of the aerosol?

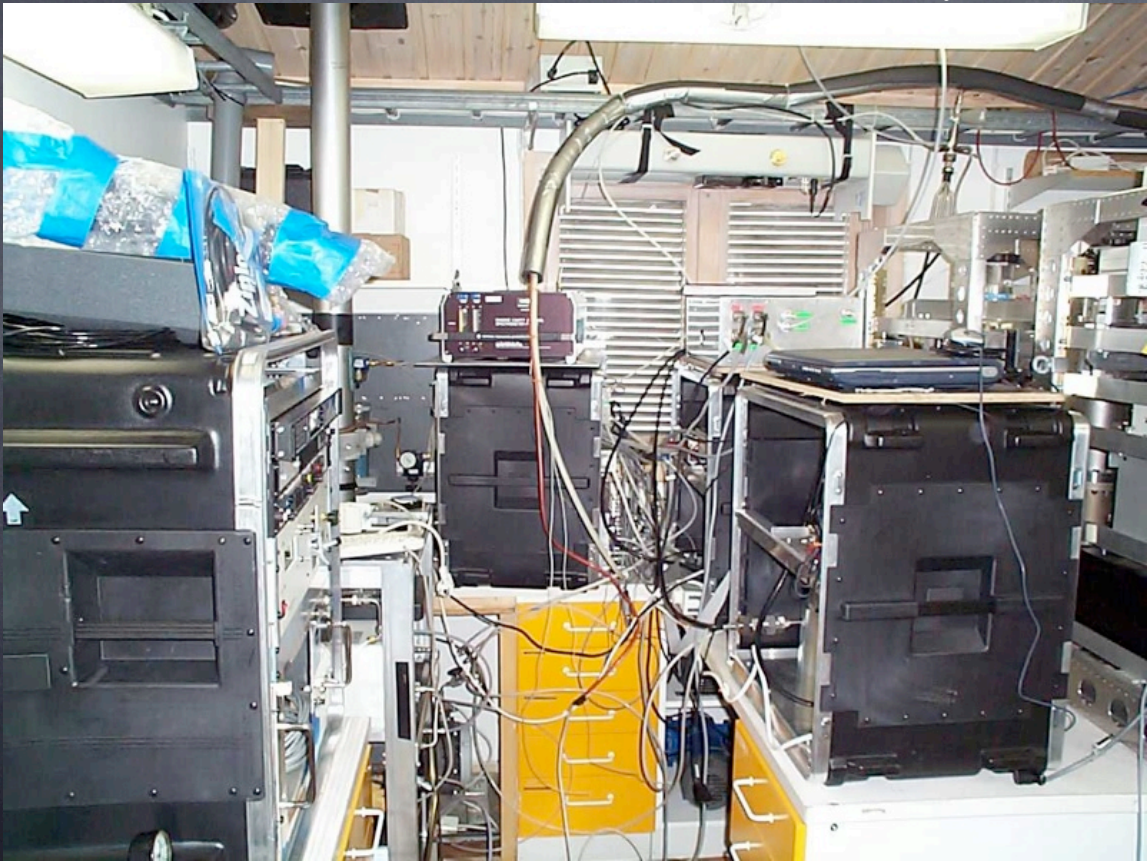
What are the chemical and physical properties of the ice residuals?

SOACED Instrumentation



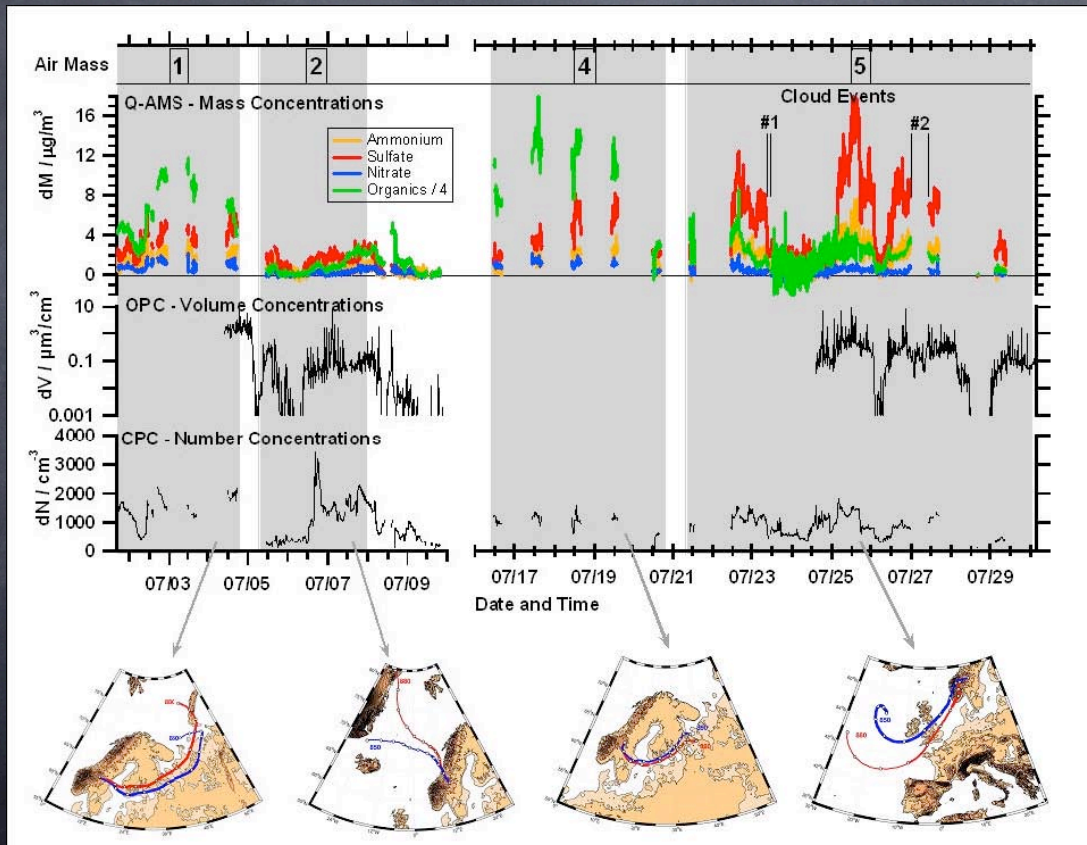
Targino, et al., Atmos. Res. 86, 225-240, 2007

Ground-based example

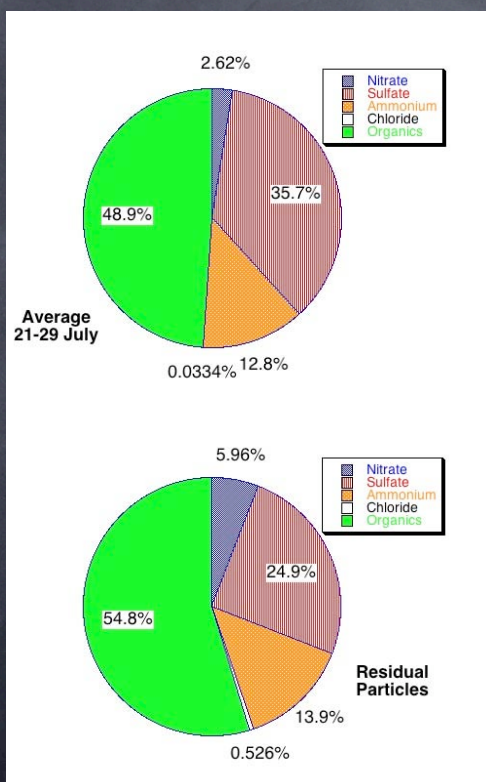


Overall chemical composition

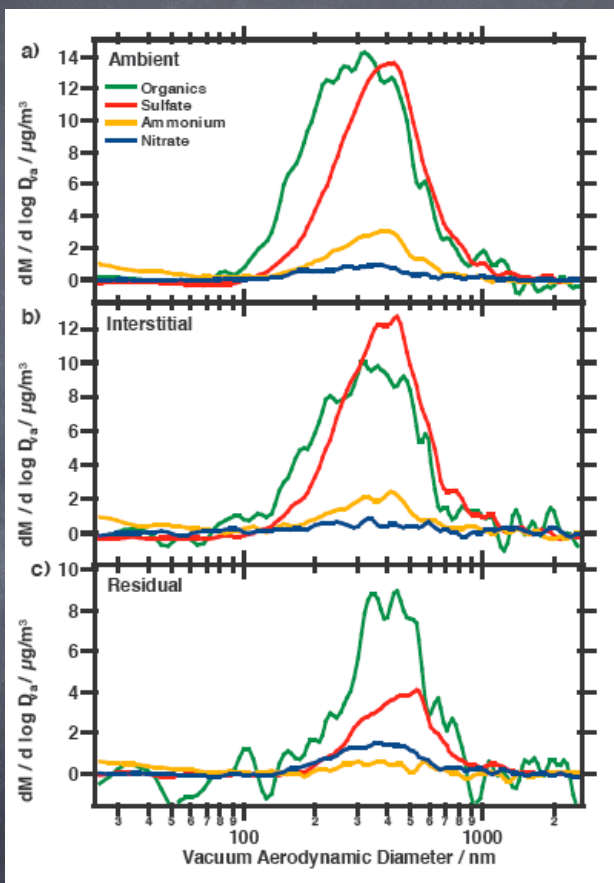
Drewnick et al., J. Atm. Chem., 56, 1-20, 2007



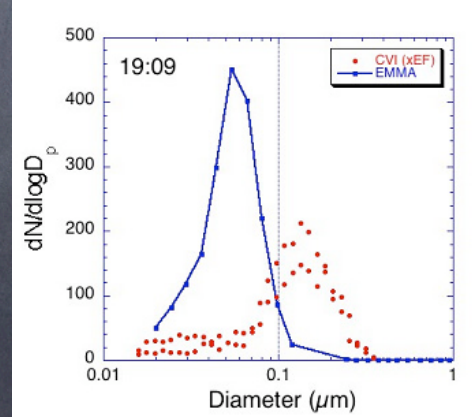
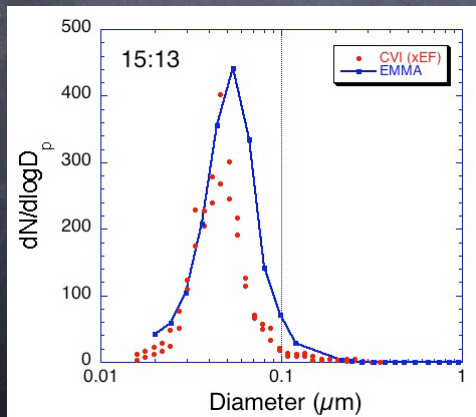
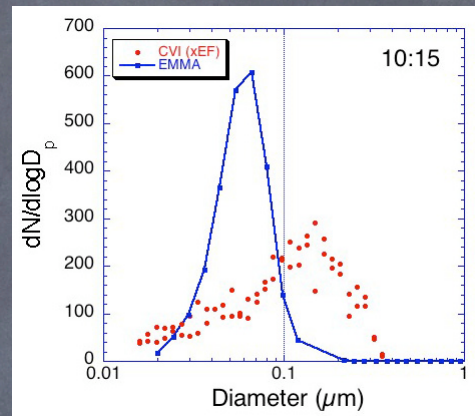
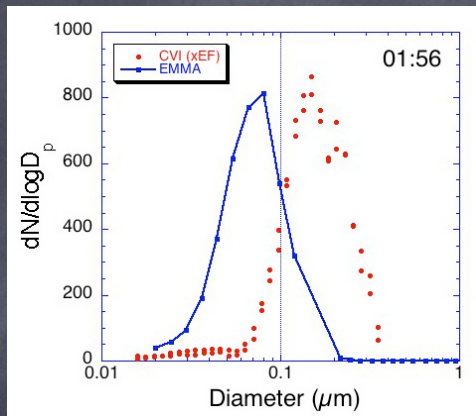
AMS Data



Drewnick et al., J. Atm. Chem., 56, 1-20, 2007



Residual & interstitial dists.



Targino et al., *Atm. Res.*, 86, 225-240, 2007

SOACED conclusions

- Accumulation-mode particles don't always control droplet number
- Organic aerosols were a substantial fraction of the aerosol that formed cloud droplets

Acknowledgments



SOACED: Admir Targino,
Dave Covert, Lynn Russell,
John Ogren, Stephan
Borrmann, Frank Drewnick



INTACC: Paul Glantz,
Admir Targino, Paul Field,
and the MRF scientists,
technicians and crew

CVI Remarks

- Aerodynamically separates and samples cloud droplets and ice crystals; excludes small aerosol particles - excellent for looking at aerosol-cloud interactions
- A multitude of measurement techniques can be used with the CVI (even gases like H_2O_2)
- The combination of an interstitial sampling system and a CVI can produce real-time chemical and microphysical information about which particles do and do not form cloud droplets and crystals
- Possible artifacts due to droplet/crystal shattering and perhaps etching of the interior surfaces of the probe

What experiments would you like to do using a CVI?

A menagerie of flying machines

Some of the aircraft that have carried a CVI



UW C-131

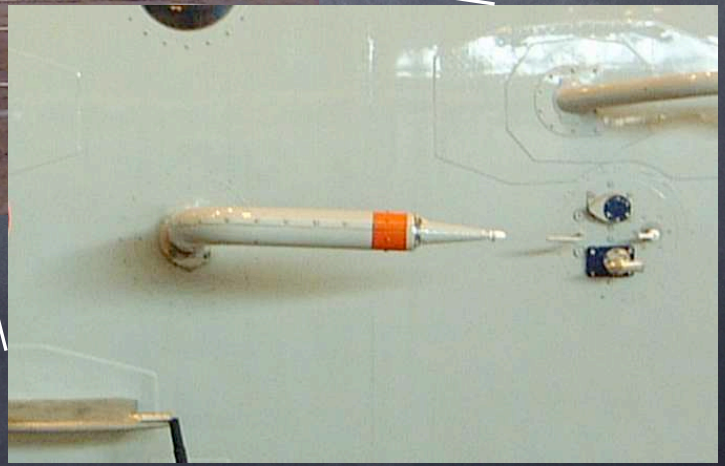
The wonderful things you can do with duct tape



DLR Falcon 20



UK C-130 (Snoopy)



NASA DC-8



Probe: Cynthia Twohy, OSU/NCAR

AES Convair 580



Battelle G-1



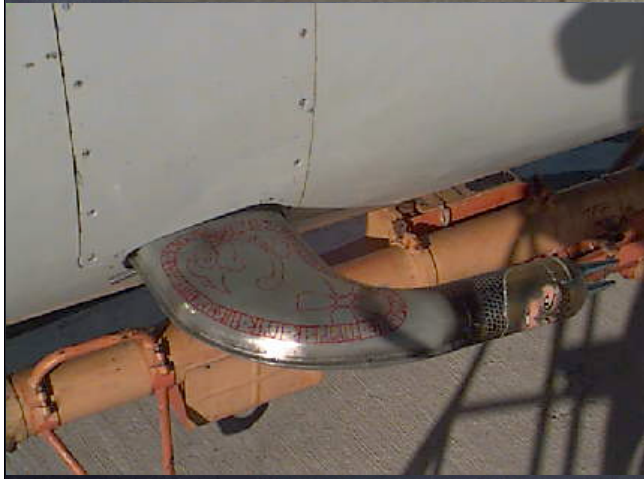
Photos: John Ogren,
NOAA

Cessna ??



Photo: Stephan Mertes, IFT

M55 Geophysika



Inlets - Citation



- Inlets take up the emergency exit hatch (!)
- Complicated airflow
- CVI

Photo: Johan Ström

Thanks for your attention!

